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Preparation of Tetrasubstituted Olefins Using Mono or Double Aerobic Direct C-H Functionalization Strategies: Importance of **Steric Effects**

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Supporting Information

ABSTRACT: A novel protocol for the synthesis of tetrasubstituted olefins through a biomimetic approach has been explored. Both mono- and diarylations were performed under ambient oxygen pressure, giving a range of highly hindered tetrasubstituted alkenes. For diarylation of disubstituted substrates, it was demonstrated that the second arylation is the rate-limiting step of the overall transformation.

■ INTRODUCTION

Tetrasubstituted olefins constitute an important class of compounds since many of these olefins show significant biological activities (Figure 1). For example, (Z)-Tamoxifen displays effects against breast cancer¹ while Rofecoxib is a powerful nonsteroidal anti-inflammatory drug.² Dibenzoxapin and related compounds have been evaluated as nuclear hormone receptor modulators,³ and finally, tetrasubstituted isocombretastatins A-4 have been recently identified as new tubulin inhibitors.4

Reported efficient methods for accessing such unsaturated structures are mainly based on the use of transition-metal catalysis via carbofunctionalization of alkynes,⁵ olefin metathesis, or cross-coupling reactions. Among the latter, the oxidative Heck coupling has been frequently employed for the preparation of disubstituted alkenes.8 However, only a few examples of successful Heck arylation have been reported regarding the synthesis of tri- or tetrasubstituted olefins. 7b-e There are several problems associated with the oxidative Heck coupling between aromatic heterocycles and trisubstituted olefins to give tetrasubstituted olefins. The problems with the latter reaction can be rationalized by the low reactivity of the trisubstituted substrates. Due to steric hindrance around the unsaturated core, the latter alkenes are not reactive enough to undergo the required carbopalladation. Another problem is associated with the regeneration of the active catalyst. In general, the use of strong oxidants or additives-such as TEMPO (2,2,6,6-tetramethylpiperidine-N-oxyl radical) derivatives e and/ or inorganic salts 7c,d—is required, thus reducing the applicability and sustainability of the reaction. In addition, in the dehydrogenative version of the Heck reaction—the Fujiwara-Moritani reaction⁹—the challenging metal insertion into the aromatic C— H bond makes the synthetic task even more difficult. Therefore, there is a demand for improvements of the synthesis of tetrasubstituted alkenes via the dehydrogenative Heck reaction approach.

In the past few years, we have been involved in the development of new sustainable C-C couplings via C-H activations using a biomimetic approach.¹⁰ Following this concept, the high kinetic barrier preventing the catalyst regeneration is circumvented by the use of catalytic amounts of electron-transfer mediators (ETMs). ¹¹ In this way, the reduced catalyst can be reoxidized by O2 at atmospheric pressure, producing water as the sole byproduct of the reaction. On the basis of this strategy, we have previously established protocols for the dehydrogenative Heck reaction that have the following advantages: (i) relative low palladium and arene loadings, (ii) utilization of O₂ under ambient pressure as the oxidant, and (iii) extension of the scope to nonbiased olefins and heterocycles. Our continued interest in this field prompted us to explore the preparation of tetrasubstituted olefins via a biomimetic approach, and our contribution is reported herein.

RESULTS AND DISCUSSION

We first planned to prepare a trisubstituted olefin via a dehydrogenative Heck reaction that could be used as starting material for the synthesis of tetrasubstituted olefins. In our previously reported arylation of nonbiased olefins, 10a we showed that acridine as a ligand dramatically enhances the reaction rate and totally controls the site selectivity in the coupling with veratrole. We initiated our studies with a 1:10 ratio of alkene 1a and veratrole (2a) using Pd(OAc)₂ (5 mol %) as catalyst, acridine (5 mol %) as ligand, and benzoquinone (BQ) (10 mol %) and iron phthalocyanine Fe(Pc) (2.5 mol %) as electrontransfer mediators in a mixture of acetic acid:dioxane (1:1, v:v)

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The Journal of Organic Chemistry

Figure 1. Representative drugs containing tetrasubstituted olefins.

Table 1. Optimization of the Synthesis of Tetrasubstituted Olefins^a

	conditions				
entry	T (°C)	arene (equiv)	variations from standard conditions (concerning solvent and catalysts)	3aa (%) ^b	4aa $(\%)^b$
1	90	10	none	66	27
2	90	10	AcOH as solvent	59	17
3	100	10	none	36	35
4	110	10	none	42	3
5	100	10	BQ (20 mol %) and Fe(Pc) (5 mol %)	30	39
6	100	10	AcOH:dioxane (75:25) as solvent	57	29
7	100	10	AcOH:dioxane (25:75) as solvent	28	9
8	100	10	Pd(OAc) ₂ (7.5 mol %)	24	66
9	100	15	none	trace	66 ^c
10	100	15	AcOH:CH ₃ CN (8:2) as solvent	56	9
11	100	15	PivOH:dioxane (1:1) as solvent	33	7
12	100	15	$C_2H_5COOH:$ dioxane (1:1) as solvent	18	6
13	100	15	1,4-diOMeBQ (10 mol %) instead of BQ	51	37
14	100	15	Cu(Pc) (2.5 mol %) instead of Fe(Pc)	49	37

"Reaction conditions: 1a (0.30 mmol), 2a (10 or 15 equiv) in the appropriate catalytic system for 24 h under O₂ (balloon). ^bNMR yield using an internal standard. ^cYield after flash chromatography.

for 24 h at 90 $^{\circ}$ C under ambient oxygen pressure (Table 1, entry 1). Interestingly, we found that formation of the trisubstituted alkene 3aa was accompanied by the tetrasubstituted alkene 4aa. This reaction shows that the biomimetic approach is a viable strategy for providing access to tetrasubstituted olefins. Taking into account that there are not many examples in the literature for the diarylation of alkenes, ^{10f,12} it was highly interesting to develop a one-pot double arylation of 1a.

Attempts to increase the rate of the reaction by the use of pure acetic acid as the solvent were unsuccessful and led to only 17% yield of 4aa (Table 1, entry 2). An increase of the reaction temperature to 100 °C under the standard conditions resulted in an improvement and gave olefin 4aa in a 35% yield, (Table 1, entry 3). However, a further increase of the reaction temperature to 110 °C decreased the amount of 4aa to 3% (entry 4). The dramatic decrease of 4aa may be due to decomposition at 110 °C under the acidic conditions. An increase of the catalytic amount of ETMs did not significantly affect the yield of the coupling, and modifications of the solvent ratio led to decreased yields (entries 5–7). We were pleased to find that the use of a higher catalyst loading (entry 8) or an increase of the arene loading (entry 9)

improved the yield of 4aa up to 66%. Considering the importance of the choice of solvent in the Fujiwara–Moritani reaction, we also evaluated the role of a range of cosolvents such as acetonitrile instead of dioxane, ¹³ or pivalic acid or propionic acid instead of acetic acid, but none of these changes increased the yield of 4aa (entries 10–12). We chose to conclude our optimization studies with an additional screening of ETMs, but these alternative catalytic systems were not more efficient than those used in the standard conditions (entries 13–14).

The double dehydrogenative sequence for the conversion of disubstituted alkene 1a into trisubstituted alkene 3aa and tetrasubstituted alkene 4aa was monitored by ¹H NMR spectroscopy (Figure 2). The reaction profile indicates that olefin 1a is almost totally consumed after only 2 h, mostly giving 3aa with only trace amounts of 4aa. Then, the concentration of 3aa is decreasing slowly with concomitant formation of 4aa, demonstrating that the rate-limitating step of the sequence is the formation of the desired tetrasubstituted product. The steric hindrance around the double bond certainly slows down the carbopalladation.

The Journal of Organic Chemistry

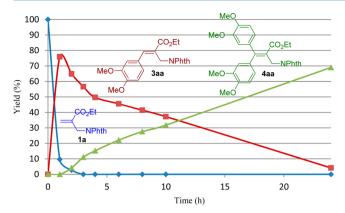
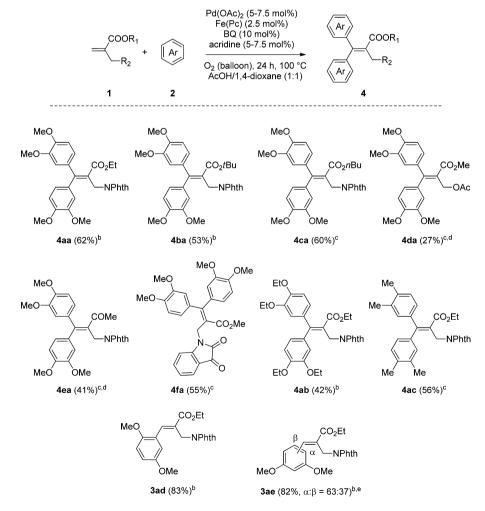


Figure 2. Reaction profile of the biomimetic double dehydrogenative sequence between alkene **1a** and arene **2a** using the reaction conditions of entry **9**. Table 1.

We chose to continue our studies with a range of one-pot diarylations using the optimum conditions (Scheme 1). In most cases, the introduction of directing groups—such as acyl groups—can only be employed to partially control the regio-and the stereoselectivity of tetrasubstituted alkenes. ^{7d,14} Indeed, simple arenes can potentially undergo metalation at several

reactive sites, generating complicated mixtures of isomers after a double cross-coupling. Usually, the site selectivity is controlled by (i) electronic factors with a preference for the most electronrich carbon, (ii) steric factors with a preference for the lesshindered carbon. First, the influence of the substituents in 1,1disubstituted alkene substrates was examined by reaction with veratrole 2a as the aromatic coupling partner. We were pleased to find that a range of esters smoothly underwent the diarylation, giving 4aa-4ca in good yields. An acetate and a ketone are both tolerated in the double dehydrogenative cross-coupling, albeit in lower yields (4da and 4ea). An olefin substrate containing an isatin moiety underwent a smooth reaction, resulting in the formation of 4fa in 55% yield. To our satisfaction, the site selectivity of the reaction with 1,2-diethoxybenzene and o-xylene was complete, leading to two highly substituted scaffolds 4ab and 4ac. However, no diarylated product was observed when 1,4dimethoxybenzene 2d was employed as coupling partner, the reaction yielding only the trisubstituted olefin 3ad in an 83% yield. The second arylation is apparently suppressed due to steric reasons. The lack of reactivity due to steric effects was confirmed by using 1,3-dimethoxybenzene 2e as coupling partner. Indeed, a 63:37 mixture in favor of the *ortho*-isomer $3ae-\alpha$ (the most reactive site) was isolated, accompanied by roughly 5% of a diarylated scaffold. In this example, due to its steric hindrance,

Scheme 1. Synthesis of Tetrasubstituted Olefins via a Double Aerobic Direct C-H Activation a,b,c,d,e



"For reaction conditions, see Table 1. ${}^b\text{Pd}(\text{OAc})_2$ (5 mol %), acridine (5 mol %). ${}^c\text{Pd}(\text{OAc})_2$ (7.5 mol %), acridine (7.5 mol %). ${}^d\text{Reaction}$ performed at 90 °C. "Ratio of isomers ($\alpha:\beta$) determined by NMR spectroscopy of isolated product.

The Journal of Organic Chemistry

Scheme 2. Synthesis of Tetrasubstituted Olefins via a Mono Aerobic Direct C-H Activation a,b,c,d,e

^aFor reaction conditions, see Table 1. ${}^{b}Pd(OAc)_{2}$ (5 mol %), acridine (5 mol %). ${}^{c}Pd(OAc)_{2}$ (7.5 mol %), acridine (7.5 mol %). ${}^{d}Ratio$ of isomers (α:β or o:m:p) determined by NMR spectroscopy of isolated product. ^cRatio of regioisomers, which could not be assigned and are thus given in no particular order (determined by NMR of isolated product).

3ae- α is already too crowded to perform a second arylation with **2e**. In addition, **3ae-** β can only react with the β position of **2e**, which is unfortunately the less reactive carbon of the arene.

We also confirmed the influence of steric hindrance starting from trisubstituted alkenes 1g and 1h (Scheme 2). Reaction of 1g with arene 2d did not deliver the desired product 5gd, the starting materials being mostly recovered and no identifiable byproduct being detected. Furthermore, with arene 2e, only one isomer 5ge was obtained in a 16% yield, whereas 57% of olefin 1g remained intact. These systematic studies on these sterically hindered substrates led to some instructive results: ortho-C-H functionalization of simple arenes is very slow with trisubstituted olefins, illustrating the influence of steric effects on the rate. The reaction was also conducted with 1,3-benzodioxole (1f) as coupling partner, and surprisingly, the selectivity of the coupling was not complete. The desired alkenes 5gf were isolated as a mixture of isomers in a 18:82 ratio in favor of the β -alkenylated scaffold. In light of these results, it was of interest to establish the selectivity of anisole as coupling partner. Anisole is known to mainly undergo palladium insertion at (i) para, (ii) ortho, (iii) meta positions. 10a-c,13As expected, no coupling occurred at the ortho position of anisole, and 5gg was isolated as a mixture of isomers in a 0:46:54 (o:m:p) ratio. Similarly, toluene and chlorobenzene were also successfully employed and only two regioisomers were detected in each case (5gh and 5gi). To

further explore the scope of this transformation, the coupling reaction with 1g (or 1h) was conducted with difunctionalized arenes such as veratrole, naphtalene, or *o*-xylene, which furnished a range of densely substituted alkenes 5ga, 5gj, 5gc, and 5ha with synthetically useful yields and complete selectivity. 15

CONCLUSION

In summary, we have developed an operationally simple protocol for the synthesis of tetrasubstituted olefins via mono or double aerobic dehydrogenative Heck couplings through a biomimetic approach. It was shown that the steric hindrance around the unsaturated core plays a key role in the selectivity of the reaction, and a range of highly substituted alkenes were isolated with complete chemoselectivity around the double bond and with partial to complete regioselectivity depending on the arene. Remarkably, the reaction involves readily available nonfunctionalized reagents and proceeds at ambient oxygen pressure.

EXPERIMENTAL SECTION

General Information. Reactions were monitored by thin-layer chromatography (TLC) analysis using silica gel (60 F254) plates. Compounds were visualized by UV irradiation and/or spraying with a solution of potassium permanganate, followed by charring at 150 °C. Flash column chromatography was performed on silica gel 60 (230–400 mesh, 0.040–0.063 mm). ¹H and ¹³C NMR spectra were recorded on a spectrometer at 400 MHz (¹³C, 100 MHz). Chemical shifts are given in

parts per million from tetramethylsilane (TMS) as internal standard. The following abbreviations are used for the proton spectra multiplicities: s: singulet, d: doublet, t: triplet, q: quartet, qu: quintuplet, sex:sextet, m: multiplet. Coupling constants (*J*) are reported in hertz (Hz). HRMS were recorded using ESI-TOF techniques. Dry solvents were obtained from a VAC solvent purifier. All reagents were obtained from commercial suppliers unless otherwise stated.

Protecting alkenes 1 were prepared following a two-step sequence Baylis—Hillman reaction (giving products 6)/Mitsunobu reaction as described below.

tert-Butyl 2-(Hydroxymethyl)acrylate (6b). The title compound was prepared via Baylis—Hillman reaction according to a literature procedure. ¹⁶ Experimental data were in accordance with those reported in the previous literature. ¹⁶ ¹H NMR (CDCl₃, 400 MHz): δ 6.15 (m, 1H), 5.74 (m, 1H), 4.28 (m, 2H), 1.51 (s, 9H). ¹³C NMR (CDCl₃, 100 MHz): δ 165.8, 140.9, 125.0, 81.5, 63.0, 28.2.

tert-Butyl 2-(Hydroxymethyl)acrylate (6c). The title compound was prepared via Baylis—Hillman reaction according to a literature procedure. ¹⁶ Experimental data were in accordance with those reported in the previous literature. ¹⁷ ¹H NMR (CDCl₃, 400 MHz): δ 6.25 (q, J = 1.3 Hz, 1H), 5.82 (q, J = 1.4 Hz, 1H), 4.33 (m, 2H), 4.20 (t, J = 6.6 Hz, 2H), 1.67 (qu, J = 7.4 Hz, 2H), 1.40 (sex, J = 7.4 Hz, 2H), 0.95 (t, J = 7.4 Hz, 3H). ¹³C NMR (CDCl₃, 100 MHz): δ 166.5, 139.6, 125.7, 64.9, 62.8, 30.7, 19.2, 13.8.

Methyl 2-(Hydroxymethyl)acrylate (6d). The title compound was prepared via Baylis—Hillman reaction according to a literature procedure. ¹⁶ Experimental data were in accordance with those reported in the previous literature. ¹⁶ ¹H NMR (CDCl₃, 400 MHz): δ 6.23 (q, J = 0.9 Hz, 1H), 5.83 (q, J = 1.3 Hz, 1H), 4.36 (m, 2H), 3.80 (s, 3H). ¹³C NMR (CDCl₃, 100 MHz): δ 166.8, 139.4, 125.9, 62.4, 52.0.

3-(Hydroxymethyl)but-3-en-2-one (6e). The title compound was prepared via a Baylis—Hillman reaction according to a literature procedure. ¹⁸ Experimental data were in accordance with those reported in the previous literature. ¹⁸ ¹H NMR (CDCl₃, 400 MHz): δ 6.11 (s, 1H), 6.03 (t, J = 1.4 Hz, 1H), 4.29 (q, J = 0.8 Hz, 2H), 2.35 (s, 3H). ¹³C NMR (CDCl₃, 100 MHz): δ 200.5, 147.3, 126.2, 62.3, 26.0.

Methyl 2-(Hydroxy(phenyl)methyl)acrylate (6g). The title compound was prepared via Baylis—Hillman reaction according to a literature procedure. ¹⁹ Experimental data were in accordance with those reported in the previous literature. ¹⁹ ¹H NMR (CDCl₃, 400 MHz): δ 7.39–7.28 (m, 5H), 6.34 (q, J = 0.8 Hz, 1H), 5.83 (t, J = 1.2 Hz, 1H), 5.57 (s, 1H), 3.73 (s, 3H). ¹³C NMR (CDCl₃, 100 MHz): δ 166.9, 142.1, 141.4, 128.6, 128.0, 126.7, 126.3, 73.4, 52.1.

Methyl 2-((4-Chlorophenyl)(hydroxy)methyl)acrylate (6h). The title compound was prepared via Baylis—Hillman reaction according to a literature procedure. Experimental data were in accordance with those reported in the previous literature. H NMR (CDCl₃, 400 MHz): δ 7.30 (m, 4H), δ .33 (t, J = 0.8 Hz, 1H), 5.83 (t, J = 1.2 Hz, 1H), 5.51 (m, 1H), 3.70 (s, 3H). CNMR (CDCl₃, 100 MHz): δ 166.7, 141.7, 139.9, 133.6, 128.6, 128.1, 126.4, 72.7, 52.1.

Ethyl 2-((1,3-Dioxoisoindolin-2-yl)methyl)acrylate (1a). Compound 1a was prepared via a Mitsunobu reaction according to a literature procedure. ²⁰ Experimental data were in accordance with those reported in the previous literature. ²¹ H NMR (CDCl₃, 400 MHz): δ 7.88 (dd, J = 5.5, 3.1 Hz, 2H), 7.74 (dd, J = 5.6, 3.1 Hz, 2H), 6.33 (t, J = 1.4 Hz, 1H), 5.57 (t, J = 3.3 Hz, 1H), 4.57 (t, J = 3.1 Hz, 2H), 4.25 (q, J = 7.1 Hz, 2H), 1.31 (t, J = 7.2 Hz, 3H). ¹³C NMR (CDCl₃, 100 MHz): δ 167.8, 165.4, 134.7, 134.2, 132.1, 125.9, 123.5, 61.2, 38.4, 14.2.

tert-Butyl 2-((1,3-Dioxoisoindolin-2-yl)methyl)acrylate (1b). Compound 1b was prepared via a Mitsunobu reaction according to a literature procedure. ²⁰ Experimental data were in accordance with those reported in the previous literature. ²⁰ ¹H NMR (CDCl₃, 400 MHz): δ 7.88 (dd, J = 5.6, 3.1 Hz, 2H), 7.74 (dd, J = 5.5, 3.0 Hz, 2H), 6.21 (td, J = 1.4, 0.5 Hz, 1H), 5.44 (t, J = 1.7, 0.5 Hz, 1H), 4.52 (t, J = 1.6 Hz, 2H), 1.50 (s, 9H). ¹³C NMR (CDCl₃, 100 MHz): δ 167.9, 164.6, 136.1, 134.2, 132.1, 124.6, 123.6, 81.6, 38.5, 28.2.

Butyl 2-((1,3-Dioxoisoindolin-2-yl)methyl)acrylate (1c). Compound 1c was prepared via a Mitsunobu reaction according to a literature procedure. ²⁰ ¹H NMR (CDCl₃, 400 MHz): δ 7.87 (dd, J = 5.6, 3.1 Hz, 2H), 7.74 (dd, J = 5.5, 3.0 Hz, 2H), 6.31 (td, J = 1.4, 0.4 Hz, 1H),

5.56 (t, J = 1.7, 0.4 Hz, 1H), 4.56 (t, J = 1.5 Hz, 2H), 4.18 (t, J = 6.7 Hz, 2H), 1.69–1.62 (m, 2H), 1.44–1.34 (m, 2H), 0.93 (t, J = 7.4 Hz, 3H). 13 C NMR (CDCl₃, 100 MHz): δ 167.8, 165.5, 134.8, 134.3, 132.1, 125.9, 123.6, 65.1, 38.4, 30.7, 19.3, 13.8.

Methyl 2-(Acetoxymethyl)acrylate (1d). Compound **1d** was prepared according to a literature procedure. ²² ¹H NMR (CDCl₃, 400 MHz): δ 6.34 (q, J = 1.3 Hz, 1H), 5.83 (q, J = 1.3 Hz, 1H), 4.79 (t, J = 1.5 Hz, 2H), 3.77 (s, 3H), 2.08 (s, 3H). ¹³C NMR (CDCl₃, 100 MHz): δ 170.4, 165.7, 135.3, 127.6, 62.5, 52.1, 20.9.

2-(2-Methylene-3-oxobutyl)isoindoline-1,3-dione (1e). Compound **1e** was prepared via a Mitsunobu reaction according to a literature procedure. Experimental data were in accordance with those reported in the previous literature. ¹H NMR (CDCl₃, 400 MHz): δ 7.87 (dd, J = 5.6, 3.1 Hz, 2H), 7.74 (dd, J = 5.5, 3.1 Hz, 2H), 6.13 (m, 1H), 5.70 (m, 1H), 4.54 (t, J = 1.5 Hz, 2H), 2.37 (s, 3H). ¹³C NMR (CDCl₃, 100 MHz): δ 198.2, 168.0, 142.6, 134.4, 134.3, 123.7, 123.6, 37.7, 26.0.

Methyl 2-((2,3-Dioxoindolin-1-yl)methyl)acrylate (1f). To a solution of (hydroxymethyl)acrylate 6d (600 mg, 5.17 mmol, 1 equiv) in diethyl ether (25 mL) was added dropwise phosphorus tribromide (535 μL, 5.68 mmol, 1.1 equiv) at 0 °C under argon. After 1 h at 25 °C, NaHCO₃ was added and the reaction mixture was extracted with diethyl ether $(3 \times 30 \text{ mL})$. The combined organic layers were washed with brine (50 mL) and dried over MgSO₄, filtered, and concentrated under reduced pressure. The resulting allylic bromide (270 mg, 1.51 mmol, equiv) was dissolved in acetonitrile (15 mL) in the presence of indoline-2,3-dione (260 mg, 1.81 mol, 1.2 equiv) and K₂CO₃ (250 mg, 1.81 mmol, 1.2 equiv). The resulting solution was stirred for 20 h at 25 °C. H₂O was then added, and the mixture was extracted with ethyl acetate (3 \times 30 mL). The organic phase was washed with brine, dried over MgSO₄, filtered, and concentrated under reduced pressure. The desired product 1f was purified by flash chromatography (petroleum ether/ethyl acetate = 6:4 to 5:5) and isolated as an orange solid in 26% yield over two steps (327 mg). 1 H NMR (CDCl₃, 400 MHz): δ 7.58 (ddd, J = 7.5, 1.3, 0.6 Hz, 1H), 7.54 (dd, J = 7.8, 1.4 Hz, 1H), 7.11 (td, J = 7.5, 0.7 Hz, 1H), 6.89 (d, J = 7.9 Hz, 1H), 6.35 (t, J = 1.3 Hz, 1H), 5.73 (t, J = 1.5 Hz, 1H), 4.58 (s, 2H), 3.79 (s, 3H). 13 C NMR (CDCl₃, 100 MHz): δ 182.9, 165.9, 158.3, 150.5, 138.6, 133.1, 127.5, 125.5, 124.1, 117.7, 111.1, 52.4, 40.7. HRMS (ESI) m/z: [M + Na]⁺ calcd for C₁₃H₁₁NNaO₄ 268.0586, found 268.0589.

(*E*)-Methyl 2-((1,3-Dioxoisoindolin-2-yl)methyl)-3-phenyl Acrylate (1g). Compound 1g was prepared via a Mitsunobu reaction according to a literature procedure. Experimental data were in accordance with those reported in the previous literature. Experimental data were in accordance with those reported in the previous literature. H NMR (CDCl₃, 400 MHz): δ 7.95 (s, 1H), 7.78 (dd, J = 5.5, 3.1 Hz, 2H), 7.67 (dd, J = 5.5, 3.0 Hz, 2H), 7.46 (d, J = 7.7 Hz, 2H), 7.35 (t, J = 7.5 Hz, 2H), 7.27 (t, J = 7.5 Hz, 1H), 4.75 (s, 2H), 3.75 (s, 3H). NMR (CDCl₃, 100 MHz): δ 168.0, 167.1, 143.4, 134.8, 133.9, 132.1, 129.1, 128.8, 128.6, 126.5, 123.5, 52.1, 35.9.

(*E*)-Methyl 3-(4-Chlorophenyl)-2-((1,3-dioxoisoindolin-2-yl)-methyl)acrylate (1h). Compound 1h was prepared via a Mitsunobu reaction according to a literature procedure. ²⁰ Experimental data were in accordance with those reported in the previous literature. ²¹ H NMR (CDCl₃, 400 MHz): δ 7.90 (s, 1H) 7.81 (dd, J = 5.6, 3.1 Hz, 2H), 7.71 (dd, J = 5.5, 3.0 Hz, 2H), 7.45 (d, J = 8.1 Hz, 2H), 7.35 (d, J = 8.0 Hz, 2H), 4.74 (d, J = 1.1 Hz, 2H), 3.78 (s, 3H). ¹³C NMR (CDCl₃, 100 MHz): δ 168.0, 166.9, 142.0, 134.1, 133.2, 132.0, 130.4, 128.9, 127.1, 123.7, 123.3, 52.4, 35.8.

General Procedure for the Synthesis of Functionalized Olefins 3, 4, or 5. $Pd(OAc)_2$ (5 mol %), acridine (5 mol %), p-benzoquinone (10 mol %), iron phtalocyanine (2.5 mol %), olefin 1 (1 equiv), arene 2 (15 equiv), and AcOH:dioxane (1:1, 1.0 mL) were charged in a Schlenk tube. The resulting mixture was degassed three times under reduced pressure before introducing oxygen gas with a balloon. After vigorous stirring at 100 °C for 24 h, the reaction mixture was cooled to room temperature, diluted with AcOEt, filtered through Celite, rinsed with AcOEt, and concentrated under vacuum. Products were purified by flash chromatography with hexane/ethyl acetate to yield the desired functionalized olefins 3, 4, or 5.

Ethyl 3,3-Bis(3,4-dimethoxyphenyl)-2-((1,3-dioxoisoindolin2-yl)methyl)acrylate (4aa). Prepared following the general procedure. Compound 4aa was obtained as a red solid in 62% yield (66 mg) after flash chromatography (SiO₂, petroleum ether/ethyl acetate, 7:3 to 5:5). 1 H NMR (CDCl₃, 400 MHz): δ 7.79 (dd, J = 5.6, 3.0 Hz, 2H), 7.67 (dd, J = 5.6, 3.0 Hz, 2H), 7.04 (d, J = 1.8 Hz, 1H), 6.87–6.80 (m, 2H), 6.73 (d, J = 0.9 Hz, 2H), 6.67 (s, 1H), 4.68 (s, 2H), 3.88 (q, J = 7.2 Hz, 2H), 3.85 (s, 6H), 3.84 (s, 3H), 3.75 (s, 3H), 0.85 (t, J = 7.1 Hz, 3H). 13 C NMR (CDCl₃, 100 MHz): δ 169.2, 167.9, 149.1, 149.0, 148.9, 148.7, 148.4, 134.6, 133.9, 132.5, 132.1, 125.5, 123.2, 122.4, 121.7, 112.8, 112.1, 110.9, 110.5, 60.8, 56.1, 55.9, 55.9, 55.9, 39.4, 13.7. HRMS (ESI) m/z: [M + Na]⁺ calcd for C_{30} H₂₉NNaO₈ 554.1785, found 554.1782 (0.6 ppm).

(*E*)-Ethyl 3-(3,4-Dimethoxyphenyl)-2-((1,3-dioxoisoindolin-2-yl)methyl)acrylate (3aa). ¹H NMR (CDCl₃, 400 MHz): δ 7.88 (s, 1H), 7.77 (dd, J = 5.6, 3.1 Hz, 2H), 7.66 (dd, J = 5.5, 3.2 Hz, 2H), 7.14 (d, J = 1.9 Hz, 1H), 7.06 (ddd, J = 8.3, 2.0, 0.7 Hz, 1H), 7.14 (d, J = 8.4 Hz, 1H), 4.79 (d, J = 1.1 Hz, 2H), 4.18 (q, J = 7.1 Hz, 2H), 3.88 (s, 3H), 3.84 (s, 3H), 1.23 (t, J = 7.0 Hz, 3H). ¹³C NMR (CDCl₃, 100 MHz): δ 168.0, 166.9, 149.6, 148.9, 143.0, 133.9, 132.1, 127.5, 125.2, 123.2, 122.7, 112.4, 111.0, 61.1, 56.0, 55.9, 36.0, 14.2. HRMS (ESI) m/z: [M + Na]⁺ calcd for C₂₂H₂₁NNaO₆ 418.1261, found 418.1241 (4.8 ppm).

tert-Butyl 3,3-Bis(3,4-dimethoxyphenyl)-2-((1,3-dioxoiso-indolin-2-yl)methyl)acrylate (4ba). Prepared following the general procedure. Compound 4ba was obtained as a red solid in 53% yield (59 mg) after flash chromatography (SiO₂, petroleum ether/ethyl acetate, 7:3 to 6:4). ¹H NMR (CDCl₃, 400 MHz): δ 7.82 (dd, *J* = 5.5, 3.0 Hz, 2H), 7.70 (dd, *J* = 5.5, 3.1 Hz, 2H), 7.11 (d, *J* = 1.9 Hz, 1H), 6.89 (dd, *J* = 8.2, 2.0 Hz, 1H), 6.84–6.75 (m, 3H), 6.70 (d, *J* = 1.8 Hz, 1H), 4.68 (s, 2H), 3.89 (s, 3H), 3.87 (s, 3H), 3.86 (s, 3H), 3.79 (s, 3H), 1.09 (s, 9H). ¹³C NMR (CDCl₃, 100 MHz): δ 167.9, 167.8, 148.8, 148.6, 148.6, 148.3, 147.0, 134.8, 133.9, 132.6, 132.1, 127.3, 123.1, 122.2, 121.7, 112.6, 112.2, 101.8, 110.4, 81.1, 56.0, 55.9, 55.8, 55.8, 39.3, 27.4. HRMS (ESI) *m/z*: [M + Na]⁺ calcd for C₃₂H₃₃NNaO₈ 582.2098, found 582.2122 (–4.0 ppm).

Butyl 3,3-Bis(3,4-dimethoxyphenyl)-2-((1,3-dioxoisoindolin-2-yl)methyl)acrylate (4ca). Prepared following the general procedure, except that Pd(OAc)₂ (7.5 mol %) and acridine (7.5 mol %) were used. Compound 4ca was obtained as a red solid in 60% yield (67 mg) after flash chromatography (SiO₂, petroleum ether/ethyl acetate, 7:3 to 5:5). 1 H NMR (CDCl₃, 400 MHz): δ 7.80 (dd, J = 5.6, 3.1 Hz, 2H), 7.68 (dd, J = 5.5, 3.0 Hz, 2H), 7.03 (d, J = 1.8 Hz, 1H), 6.86 (dd, J = 8.2, 1.9 Hz, 1H), 6.81 (d, J = 8.3 Hz, 1H), 6.74 (m, 2H), 6.67 (s, 1H), 4.68 (s, 2H), 3.86 (s, 3H), 3.86 (s, 3H), 3.85 (s, 3H), 3.81 (t, J = 6.6 Hz, 2H), 3.76 (s, 3H), 1.19 (m, 2H), 0.96 (m, 2H), 0.67 (t, J = 7.3 Hz, 3H). 13 C NMR (CDCl₃, 100 MHz): δ 169.5, 167.9, 149.1, 149.0, 148.9, 148.7, 148.5, 134.7, 134.0, 132.5, 132.2, 125.5, 123.3, 122.5, 121.8, 112.8, 112.1, 110.9, 110.5, 64.9, 56.1, 55.9, 55.9, 55.9, 39.5, 30.3, 18.9, 13.6. HRMS (ESI) m/z: [M + Na]⁺ calcd for $C_{32}H_{33}NNaO_8$ 582.2098, found 582.2093 (0.8 ppm).

Methyl 2-(Acetoxymethyl)-3,3-bis(3,4-dimethoxyphenyl)-acrylate (4da). Prepared following the general procedure, except that Pd(OAc)₂ (7.5 mol %) and acridine (7.5 mol %) at 90 °C were used. Compound 4da was obtained as a red solid in 27% yield (23 mg) after flash chromatography (SiO₂, petroleum ether/ethyl acetate, 6:4 to 4:6). ¹H NMR (CDCl₃, 400 MHz): δ 6.83–6.71 (m, SH), 6.63 (d, J = 2.0 Hz, 1H), 4.85 (s, 2H), 3.89 (s, 3H), 3.88 (s, 3H), 3.79 (s, 3H), 3.77 (s, 3H), 3.54 (s, 3H), 2.09 (s, 3H). ¹³C NMR (CDCl₃, 100 MHz): δ 170.7, 170.3, 153.4, 150.0, 150.0, 148.6, 148.6, 134.2, 132.3, 125.0, 123.0, 122.1, 113.1, 112.3, 110.8, 110.5, 60.0, 56.0, 56.0, 56.0, 55.9, 52.0, 21.1. HRMS (ESI) m/z: [M + Na]⁺ calcd for C₂₃H₂₆NaO₈ 453.1520, found 453.1524 (-0.9 ppm).

2-(2-(Bis(3,4-dimethoxyphenyl))methylene)-3-oxobutyl)-isoindoline-1,3-dione (4ea). Prepared following the general procedure, except that $Pd(OAc)_2$ (7.5 mol %) and acridine (7.5 mol %) at 90 °C were used. Compound **4ea** was obtained as a brown solid in 41% yield (41 mg) after flash chromatography (SiO₂, petroleum ether/ethyl acetate, 5:5 to 4:6). ¹H NMR (CDCl₃, 400 MHz): δ 7.77 (dd, J = 5.5, 3.1 Hz, 2H), 7.67 (dd, J = 5.5, 2.9 Hz, 2H), 7.01 (d, J = 2.0 Hz, 1H), 6.84 (dd, J = 8.2, 2.0 Hz, 1H), 6.78 (m, 2H), 6.73 (dd, J = 8.2, 2.0 Hz,

1H), 6.60 (d, J = 2.0 Hz, 1H), 4.66 (s, 2H), 3.88 (s, 3H), 3.85 (s, 3H), 3.80 (s, 3H), 3.76 (s, 3H), 1.77 (s, 3H). 13 C NMR (CDCl₃, 100 MHz): δ 205.3, 168.2, 150.1, 149.1, 149.0, 148.9, 148.8, 134.4, 134.1, 134.0, 132.9, 132.1, 123.5, 123.3, 122.7, 112.9, 112.8, 111.0, 110.8, 56.1, 56.1, 56.0, 56.0, 40.4, 30.6. HRMS (ESI) m/z: [M + Na]⁺ calcd for C₂₉H₂₇NNaO₇ 524.1680, found 524.1682 (-0.5 ppm).

Methyl 3,3-Bis(3,4-dimethoxyphenyl)-2-((2,3-dioxoindolin1-yl)methyl)acrylate (4fa). Prepared following the general procedure, except that Pd(OAc)₂ (7.5 mol %) and acridine (7.5 mol %) were used. Compound 4fa was obtained as a red solid in 55% yield (57 mg) after flash chromatography (SiO₂, petroleum ether/ethyl acetate, 6:4 to 4:6). ¹H NMR (CDCl₃, 400 MHz): δ 7.53–7.49 (m, 2H), 7.06 (td, J = 7.5, 0.8 Hz, 1H), 6.92 (m, 1H), 6.80 (d, J = 8.2 Hz, 1H), 6.73 (d, J = 8.4 Hz, 1H), 6.70–6.76 (m, 3H), 6.55 (d, J = 2.0 Hz, 1H), 4.74 (s, 2H), 3.88 (s, 3H), 3.85 (s, 3H), 3.75 (s, 3H), 3.73 (s, 3H), 3.44 (s, 3H). ¹³C NMR (CDCl₃, 100 MHz): δ 183.0, 170.2, 157.9, 150.8, 150.8, 149.6, 149.5, 148.9, 148.6, 137.9. 134.2, 131.7, 125.1, 123.6, 123.1, 122.2, 121.7, 117.9, 112.3, 112.0, 111.3, 111.2, 110.5, 56.1, 56.1, 55.9, 55.9, 52.3, 41.2. HRMS (ESI) m/z: [M + Na]⁺ calcd for C₂₉H₂₇NNaO₈ 540.1629, found 540.1637 (–1.5 ppm).

Ethyl 3,3-Bis(3,4-diethoxyphenyl)-2-((1,3-dioxoisoindolin-2-yl)methyl)acrylate (4ab). Prepared following the general procedure. Compound 4ab was obtained as a yellow solid in 42% yield (49 mg) after flash chromatography (SiO₂, petroleum ether/ethyl acetate, 7:3 to 5:5). 1 H NMR (CDCl₃, 400 MHz): δ 7.80 (dd, J = 5.5, 3.1 Hz, 2H), 7.67 (dd, J = 5.4, 3.1 Hz, 2H), 6.96 (d, J = 1.7 Hz, 1H), 6.83 (m, 2H), 6.71 (m, 2H), 6.65 (d, J = 1.8 Hz, 1H), 4.67 (s, 2H), 4.10–4.01 (m, 6H), 3.94 (q, J = 7.0 Hz, 2H), 3.85 (q, J = 7.1 Hz, 2H), 1.45–1.34 (m, 12H), 0.83 (t, J = 7.1 Hz, 3H). 13 C NMR (CDCl₃, 100 MHz): δ 169.4, 168.0, 149.5, 148.8, 148.7, 148.3, 148.1, 134.8, 133.9, 132.6, 132.2, 125.0, 123.3, 122.6, 121.9, 114.9, 114.3, 112.8, 112.5, 64.6, 64.6, 64.5, 64.5, 60.8, 39.5, 14.9, 14.9, 14.9, 14.8, 13.7. HRMS (ESI) m/z: [M + Na]⁺ calcd for C₃₄H₃₇NNaO₈ 610.2411, found 610.2438 (–4.4 ppm).

Ethyl 3,3-Bis(3,4-dimethylphenyl)-2-((1,3-dioxoisoindolin-2-yl)methyl)acrylate (4ac). Prepared following the general procedure, except that Pd(OAc)₂ (7.5 mol %) and acridine (7.5 mol %) were used. Compound 4ac was obtained as a yellow solid in 56% yield (52 mg) after flash chromatography (SiO₂, petroleum ether/ethyl acetate, 6:4). ¹H NMR (CDCl₃, 400 MHz): δ 7.80 (dd, J = 5.5, 3.0 Hz, 2H), 7.66 (dd, J = 5.5, 3.0 Hz, 2H), 7.07 (m, 3H), 6.99 (d, J = 7.8 Hz, 1H), 6.96 (s, 1H), 6.88 (dd, J = 7.7, 1.9 Hz, 1H), 4.64 (s, 2H), 3.87 (q, J = 7.1 Hz, 2H), 2.21 (s, 6H), 2.20 (s, 3H), 2.17 (s, 3H), 0.81 (t, J = 7.2 Hz, 3H). ¹³C NMR (CDCl₃, 100 MHz): δ 169.1, 167.9, 150.6, 139.6, 137.6, 136.6, 136.5, 136.4, 136.0, 133.9, 132.3, 130.5, 129.8, 129.6, 129.3, 126.9, 126.2, 125.3, 123.2, 60.7, 39.1, 19.8, 19.8, 19.7, 19.6, 13.5. HRMS (ESI) m/z: [M + Na]+ calcd for C₃₀H₂₉NNaO₄ 490.1989, found 490.1994 (-1.0 ppm).

(*E*)-Ethyl 3-(2,5-Dimethoxyphenyl)-2-((1,3-dioxoisoindolin-2-yl)methyl)acrylate (3ad). Prepared following the general procedure. Compound 3ad was obtained as a brown solid in 83% yield (66 mg) after flash chromatography (SiO₂, petroleum ether/ethyl acetate, 7:3 to 6:4). 1 H NMR (CDCl₃, 400 MHz): δ 7.95 (s, 1H), 7.75 (dd, J = 5.5, 3.0 Hz, 2H), 7.65 (dd, J = 5.5, 3.2 Hz, 2H), 7.15 (m, 1H), 6.76 (m, 2H), 4.69 (d, J = 1.2 Hz, 2H), 4.17 (q, J = 7.1 Hz, 2H), 3.78 (s, 3H), 3.74 (s, 3H), 1.22 (t, J = 7.1 Hz, 3H). 13 C NMR (CDCl₃, 100 MHz): δ 167.9, 166.5, 153.3, 151.6, 139.5, 133.8, 132.2, 127.3, 124.5, 123.2, 115.4, 115.3, 111.9, 61.1, 56.2, 55.9, 36.1, 14.2. HRMS (ESI) m/z: [M + Na]⁺ calcd for $C_{22}H_{21}$ NNaO₆ 418.1261, found 418.1254 (1.7 ppm).

(E)-Ethyl 3-(3,5-Dimethoxyphenyl)-2-((1,3-dioxoisoindolin-2-yl)methyl)acrylate (3ae- β). Prepared following the general procedure. Compounds 3ae- α and 3ae- β were obtained and separated as brown solids in 82% yield (66 mg, α : β = 63:37) after flash chromatography (SiO₂, petroleum ether/ethyl acetate, 7:3 to 6:4). ¹H NMR (CDCl₃, 400 MHz): δ7.87 (s, 1H), 7.76 (dd, J = 5.7, 3.0 Hz, 2H), 7.66 (dd, J = 5.4, 3.0 Hz, 2H), 6.59 (dd, J = 2.3, 0.7 Hz, 2H), 6.28 (d, J = 2.2 Hz, 1H), 4.74 (d, J = 1.3 Hz, 2H), 4.19 (q, J = 7.1 Hz, 2H), 3.75 (s, 6H), 1.23 (t, J = 7.2 Hz, 3H). ¹³C NMR (CDCl₃, 100 MHz): δ 167.9, 166.5, 160.8, 142.8, 136.7, 134.4, 133.9, 132.1, 127.3, 123.7, 123.1, 106.6, 100.8, 62.2, 55.5, 55.5, 36.0, 14.2. HRMS (ESI) m/z: [M + Na]+ calcd for $C_{22}H_{21}NNaO_6$ 418.1261, found 418.1256 (1.1 ppm).

(*E*)-Ethyl 3-(2,4-Dimethoxyphenyl)-2-((1,3-dioxoisoindolin-2-yl)methyl)acrylate (3ae-α). ¹H NMR (CDCl₃, 400 MHz): δ 7.98 (s, 1H), 7.77 (dd, J = 5.6, 3.1 Hz, 2H), 7.65 (dd, J = 5.5, 2.9 Hz, 2H), 7.48 (dd, J = 8.4, 0.6 Hz, 1H), 6.46 (dd, J = 8.4, 2.4 Hz, 1H), 6.42 (d, J = 2.5 Hz, 1H), 4.70 (d, J = 1.1 Hz, 2H), (q, J = 7.1 Hz, 2H), 3.82 (s, 3H), 3.78 (s, 3H), 1.20 (t, J = 7.1 Hz, 3H). ¹³C NMR (CDCl₃, 100 MHz): δ 168.0, 166.9, 161.8, 158.9, 139.4, 133.8, 132.2, 131.1, 125.2, 123.1, 116.7, 104.5, 98.5, 60.9, 55.6, 55.5, 36.3, 14.2. HRMS (ESI) m/z: [M + Na]+ calcd for C₂₂H₂₁NNaO₆ 418.1261, found 418.1281 (-4.7 ppm).

(*E*)-Methyl 3-(3,5-Dimethoxyphenyl)-2-((1,3-dioxoisoindolin2-yl)methyl)-3-phenyl Acrylate (5ge). Prepared following the general procedure. Compound 5ge was obtained as an orange solid in 16% yield (15 mg) after flash chromatography (SiO₂, petroleum ether/ethyl acetate/toluene, 5:3:2). ¹H NMR (CDCl₃, 400 MHz): δ 7.80 (dd, J = 5.6, 3.1 Hz, 2H), 7.68 (dd, J = 5.5, 3.0 Hz, 2H), 7.25 (m, 3H), 7.18 (m, 2H), 6.52 (d, J = 2.3 Hz, 2H), 6.34 (t, J = 2.3 Hz, 1H), 4.66 (s, 2H), 3.74 (s, 6H), 3.39 (s, 3H). ¹³C NMR (CDCl₃, 100 MHz): δ 169.1, 167.8, 160.8, 149.8, 141.5, 141.2, 134.4, 134.0, 132.1, 128.3, 128.1, 123.7, 123.3, 107.2, 100.3, 55.5, 51.9, 39.2. HRMS (ESI) m/z: [M + Na]⁺ calcd for $C_{27}H_{23}NNaO_6$ 480.1418, found 480.1430 (-2.6 ppm).

(E)-Methyl 3-(Benzo[d][1,3]dioxol-4-yl)-2-((1,3-dioxoisoindolin-2-yl)methyl)-3-phenyl Acrylate (5gf- α) and (E)-Methyl 3-(3,4-Dimethoxyphenyl)-2-((1,3-dioxoisoindolin-2-yl)methyl)-**3-phenyl Acrylate** (**5gf-** β **).** Prepared following the general procedure. Compounds $\mathbf{5gf}$ - $\boldsymbol{\alpha}$ and $\mathbf{5gf}$ - $\boldsymbol{\beta}$ were obtained as an inseparable mixture of isomers as a red solid in 59% yield (52 mg, $\alpha:\beta=18:82$) after flash chromatography (SiO₂, petroleum ether/ethyl acetate, 75:25 to 6:4). ¹H NMR (CDCl₃, 400 MHz): δ 7.82 (m, 2.44H), 7.69 (m, 2.44H), 7.26 (m, 3.66H), 7.14 (m, 2.44H), 6.86 (dd, I = 7.8, 2.0 Hz, 1H), 6.79 (m, 2H), 6.74 (m, 0.44H), 6.65 (dd, J = 5.4, 3.7 Hz, 0.22H), 5.93 (s, 2H), 5.89 (s, 2H)0.44H), 4.67 (s, 2H), 4.62 (s, 0.44H), 3.41 (s, 0.66H), 3.36 (s, 3H). ¹³C NMR (CDCl₃, 100 MHz): δ 169.2, 169.2, 167.9, 167.8, 150.2, 147.8, 147.7, 147.6, 141.7, 140.5, 134.0, 133.9, 133.5, 132.2, 132.1, 129.2, 128.6, 128.4, 128.2, 128.2, 128.1, 128.1, 127.8, 123.9, 123.4, 123.4, 123.3, 122.6, 121.9, 121.2, 110.0, 109.8, 108.5, 108.3, 101.3, 101.2, 52.0, 51.8, 39.2, 38.7. HRMS (ESI) m/z: $[M + Na]^+$ calcd for $C_{26}H_{19}NNaO_6$ 464.1105, found 464.1102 (0.5 ppm).

(E)-Methyl 2-((1,3-Dioxoisoindolin-2-yl)methyl)-3-(3methoxyphenyl)-3-phenyl Acrylate (5gg-m) and (E)-Methyl 2-((1,3-Dioxoisoindolin-2-yl)methyl)-3-(4-methoxyphenyl)-3**phenyl Acrylate (5gg-p).** Prepared following the general procedure, except that Pd(OAc)₂ (7.5 mol %) and acridine (7.5 mol %) were used. Compounds 5gg-m and 5gg-p were obtained as an inseparable mixture of isomers as a yellow solid in 51% yield (44 mg, o:m:p = 0:46:54) after flash chromatography (SiO₂, petroleum ether/ethyl acetate, 75:25). ¹H NMR (CDCl₃, 400 MHz): δ 7.82 (m, 3.70H), 7.68 (m, 3.70H), 7.28– 7.13 (m, 12.45H), 6.98 (dd, J = 2.6, 1.5 Hz, 0.85H), 6.90 (m, 2.85H), 6.81 (ddd, *J* = 8.3, 2.6, 1.0 Hz, 0.85H), 4.67 (s, 2H), 4.64 (s, 1.7H), 3.79 (s, 3H), 3.78 (s, 2.55H), 3.39 (s, 2.55H), 3.36 (s, 3H). ¹³C NMR (CDCl₃, 100 MHz): δ 169.4, 169.1, 168.0, 167.8, 159.7, 159.6, 150.6, 150.0, 142.1, 141.4, 141.0, 134.0, 134.0, 132.2, 132.2, 132.2, 131.0, 129.6, 129.1, 128.7, 128.4, 128.1, 128.1, 128.1, 128.1, 126.4, 125.4, 123.3, 121.6, 114.6, 114.0, 113.9, 55.4, 55.3, 51.9, 51.8, 39.3, 39.2. HRMS (ESI) *m/z*: $[M + Na]^+$ calcd for $C_{26}H_{21}NNaO_5$ 450.1312, found 450.1331 (-4.9)

(E)-Methyl 2-((1,3-Dioxoisoindolin-2-yl)methyl)-3-phenyl-3-(m-tolyl)acrylate and (E)-Methyl 2-((1,3-Dioxoisoindolin-2-yl)methyl)-3-phenyl-3-(p-tolyl)acrylate (5gh). Prepared following the general procedure, except that Pd(OAc)₂ (7.5 mol %) and acridine (7.5 mol %) were used. Compounds 5gh were obtained as an inseparable mixture of isomers as a yellow solid in 57% yield (47 mg, ratio that could not be assigned and is thus given in no particular order = 0:40:60) after flash chromatography (SiO2, petroleum ether/ethyl acetate/toluene, 5:3:2). ¹H NMR (CDCl₃, 400 MHz): δ 7.80 (m, 3.4H), 7.68 (m, 3.4H), 7.27-7.07 (m, 14.3H), 4.65 (s, 1.4H), 4.63 (s, 2H), 3.39 (s, 3H), 3.38 (s, 2.1H), 2.33 (s, 2.1H), 2.30 (s, 3H). 13 C NMR (CDCl₃, 100 MHz): δ 169.3, 169.3, 167.9, 167.8, 150.7, 150.7, 141.9, 141.7, 139.7, 138.2, 138.2, 136.9, 134.0, 132.2, 132.2, 129.8, 129.4, 129.2, 129.1, 129.0, 128.6, 128.5, 128.4, 128.3, 128.1, 128.1, 126.3, 126.0, 125.7, 125.4, 123.3, 123.3, 51.9, 51.9, 39.2, 39.1, 21.5, 21.4. HRMS (ESI) m/z: $[M + Na]^+$ calcd for C₂₆H₂₁NNaO₄ 434.1363, found 434.1372 (-2.2 ppm).

(*E*)-Methyl 3-(3-Chlorophenyl)-2-((1,3-dioxoisoindolin-2-yl)-methyl)-3-phenyl Acrylate and (*E*)-Methyl 3-(4-Chlorophenyl)-2-((1,3-dioxoisoindolin-2-yl)methyl)-3-phenyl Acrylate (5gi). Prepared following the general procedure, except that Pd(OAc) $_2$ (7.5 mol %) and acridine (7.5 mol %) were used. Compounds 5gi were obtained as an inseparable mixture of isomers as a yellow solid in 36% yield (31 mg, ratio which could not be assigned and is thus given in no particular order = 0:63:37) after flash chromatography (SiO $_2$) petroleum ether/ethyl acetate/toluene, 5:3:2). ¹H NMR (CDCl $_3$, 400 MHz): δ 7.81 (m, 3.16H), 7.69 (m, 3.16H), 7.35–7.11 (m, 14.22H), 4.61 (s, 2H), 4.60 (s, 1.16H), 3.39 (s, 1.74 H), 3.38 (s, 3H). ¹³C NMR (CDCl $_3$, 100 MHz): δ 168.9, 168.8, 167.8, 167.8, 148.9, 148.5, 141.4, 141.2, 140.9, 138.1, 134.5, 134.4, 134.1, 134.1, 132.1, 132.1, 130.8, 129.9, 129.3, 128.8, 128.5, 128.5, 128.4, 128.4, 128.3, 128.3, 128.2, 127.5, 127.2, 126.7, 123.4, 123.4, 52.0, 52.0, 39.0, 39.0. HRMS (ESI) m/z: [M + Na]+ calcd for C $_2$ 5H $_1$ 8ClNNaO $_4$ 454.0817, found 454.0827 (–2.2 ppm).

(*E*)-Methyl 3-(3,4-Dimethoxyphenyl)-2-((1,3-dioxoisoindolin-2-yl)methyl)-3-phenyl Acrylate (5ga). Prepared following the general procedure. Compound 5ga was obtained as a red solid in 55% yield (50 mg) after flash chromatography (SiO₂, petroleum ether/ethyl acetate, 7:3). ¹H NMR (CDCl₃, 400 MHz): δ 7.80 (dd, J = 5.5, 3.1 Hz, 2H), 7.68 (dd, J = 5.5, 3.0 Hz, 2H), 7.25 (m, 3H), 7.15 (m, 2H), 6.99 (d, J = 1.7 Hz, 1H), 6.82 (m, 2H), 4.69 (s, 2H), 3.84 (s, 3H), 3.84 (s, 3H), 3.37 (s, 3H). ¹³C NMR (CDCl₃, 100 MHz): δ 169.3, 167.9, 150.2, 149.0, 148.7, 141.9, 134.0, 132.4, 132.1, 128.6, 128.1, 128.1, 125.8, 123.3, 122.2, 112.7, 110.9, 56.1, 55.9, 51.8, 39.4. HRMS (ESI) m/z: [M + Na]⁺ calcd for $C_{27}H_{23}NNaO_6$ 480.1418, found 480.1433 (-3.3 ppm).

(*E*)-Methyl 2-((1,3-Dioxoisoindolin-2-yl)methyl)-3-(naphthalen-2-yl)-3-phenyl Acrylate (5gj). Prepared following the general procedure, except that Pd(OAc)₂ (7.5 mol %) and acridine (7.5 mol %) were used. Compound 5gj was obtained as a brown solid in 83% yield (74 mg) after flash chromatography (SiO₂, petroleum ether/ethyl acetate/toluene, 7:3:2). ¹H NMR (CDCl₃, 400 MHz): δ 7.91 (d, J = 1.1 Hz, 1H), 7.85–7.76 (m, 5H), 7.64 (dd, J = 5.5, 3.1 Hz, 2H), 7.48 (m, 2H), 7.39 (dd, J = 8.5, 1.7 Hz, 1H), 7.28–7.16 (m, 5H), 4.70 (s, 2H), 3.43 (s, 3H). ¹³C NMR (CDCl₃, 100 MHz): δ 169.2, 167.9, 150.2, 141.6, 137.1, 134.0, 133.1, 133.0, 132.1, 128.7, 128.7, 128.4, 128.3, 128.2, 128.2, 127.8, 127.1, 126.6, 126.6, 126.5, 123.2, 51.9, 39.2. HRMS (ESI) m/z: [M + Na]⁺ calcd for C₂₉H₂₁NNaO₄ 470.1363, found 470.1343 (4.2 ppm).

(*E*)-Methyl 3-(3,4-Dimethylphenyl)-2-((1,3-dioxoisoindolin-2-yl)methyl)-3-phenyl Acrylate (5gc). Prepared following the general procedure, except that Pd(OAc) $_2$ (7.5 mol %) and acridine (7.5 mol %) were used. Compound 5gc was obtained as a yellow solid in 60% yield (51 mg) after flash chromatography (SiO $_2$, petroleum ether/ethyl acetate/toluene, 7:3:2). ¹H NMR (CDCl $_3$, 400 MHz): δ 7.80 (dd, J = 5.5, 3.0 Hz, 2H), 7.68 (dd, J = 5.5, 3.2 Hz, 2H), 7.25 (m, 3H), 7.16 (m, 2H), 7.10–7.04 (m, 3H), 4.67 (s, 2H), 3.89 (s, 3H), 2.22 (s, 3H), 2.20 (s, 3H). ¹³C NMR (CDCl $_3$, 100 MHz): δ 169.4, 167.8, 150.8, 142.0, 137.2, 136.8, 136.6, 133.9, 132.2, 130.4, 129.7, 128.5, 128.0, 127.9, 126.8, 125.5, 123.2, 51.7, 39.2, 19.8, 19.8. HRMS (ESI) m/z: [M + Na] $^+$ calcd for $C_{27}H_{23}$ NNaO $_4$ 448.1519, found 448.1537 (–4.0 ppm).

(*Z*)-Methyl 3-(4-Chlorophenyl)-3-(3,4-dimethoxyphenyl)-2-((1,3-dioxoisoindolin-2-yl)methyl)acrylate (5ha). Prepared following the general procedure. Compound 5ha was obtained as a red solid in 55% yield (54 mg) after flash chromatography (SiO₂, petroleum ether/ethyl acetate, 7:3). ¹H NMR (CDCl₃, 400 MHz): δ 7.80 (dd, J = 5.4, 2.9 Hz, 2H), 7.70 (dd, J = 5.4, 3.0 Hz, 2H), 7.23 (d, J = 8.6 Hz, 2H), 7.09 (d, J = 8.7 Hz, 2H), 6.95 (m, 1H), 6.80 (m, 2H), 4.68 (s, 2H), 3.85 (s, 3H), 3.84 (s, 3H), 3.41 (s, 3H). ¹³C NMR (CDCl₃, 100 MHz): δ 169.0, 167.9, 149.2, 149.0, 148.9, 140.4, 134.2, 134.1, 132.1, 132.0, 130.0, 128.4, 126.3, 123.3, 122.3, 112.6, 111.0, 56.1, 56.0, 52.0, 39.4. HRMS (ESI) m/z: [M + Na]⁺ calcd for C₂₇H₂₂ClNNaO₆ 514.1028, found 514.1021 (1.3 ppm).

ASSOCIATED CONTENT

S Supporting Information

Copies of 1 H and 13 C NMR spectra of alkenes 1f, 3, 4, and 5. This material is available free of charge via the Internet at http://pubs. acs.org.

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Notes

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